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Solution This problem gives us, for a hypothetical ion pair, values for r_0 (0.35 nm), E_0 6.13 eV), and n (10), and asks that we determine explicit expressions for attractive and repulsive energies of Equations 2.8 and 2.9. In essence, it is necessary to compute the values of A and B in these equations.

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Solution The average atomic weight of silicon (A) is computed by adding fraction-of-occurrence/atomic weight Si products for the three isotopes⎯i.e., using Equation 2.2. (Remember: fraction of occurrence is equal to the percent of occurrence divided by 100.) Thus. A Si = f 28 A 28 + f 29 A 29 f 30 A Si Si Si Si Si Si 30 Si

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Chapter 9: Phase Diagrams
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